# AGRICULTURAL AND FOOD CHEMISTRY

# Characterization and Fermentability of an Ethanol Soluble High Molecular Weight Coffee Fraction

DIANA GNIECHWITZ,<sup>†</sup> NICOLE REICHARDT,<sup>‡</sup> ERNST MEISS,<sup>†</sup> JOHN RALPH,<sup>\$,||</sup> HANS STEINHART,<sup>†</sup> MICHAEL BLAUT,<sup>‡</sup> AND MIRKO BUNZEL<sup>\*,⊥</sup>

Department of Food Chemistry, University of Hamburg, Grindelallee 117, 20146 Hamburg, Germany; German Institute of Human Nutrition Potsdam-Rehbruecke, Arthur-Scheunert-Allee 114–116,
14558 Nuthetal, Germany; U.S. Dairy Forage Research Center, USDA-ARS, 1925 Linden Drive West, Madison, Wisconsin 53706-1108; Biological Systems Engineering Department, University of Wisconsin, Madison, Wisconsin 53706; and Department of Food Science and Nutrition, University of Minnesota, 1334 Eckles Avenue, St. Paul, Minnesota 55108

Brews from differently roasted Arabica coffees were shown to contain 8-12% ethanol soluble substances with molecular masses greater than 2 kDa, possibly contributing to their dietary fiber contents. About 13% of these substances were nondigestible carbohydrates, mainly arabinogalactans. The nondigestible high molecular weight ethanol soluble fraction (HESF) of the medium roasted coffee brew was further characterized and subjected to in vitro fermentation with human fecal bacteria. In addition to carbohydrates, HESF contained proteins/peptides (~20%), but the main fraction was composed of structurally unknown Maillard reaction products. From NMR spectroscopy, we conclude that intact caffeic and ferulic acid derivatives were not incorporated into the melanoidins to a significant extent. Stepwise ultrafiltration and gel filtration indicated a large variation in the molecular weights of HESF constituents. Coffee HESF was shown to be less fermentable by fecal bacteria than soluble coffee fiber isolated by the enzymatic-gravimetric methodology, and because of its lower carbohydrate content, less short-chain fatty acids were produced during the fermentation. Total cell counts, destructive chemical analysis, and NMR spectroscopy indicated that coffee carbohydrates are the preferred substrates for colonic microbiota. However, NMR spectra, absorbances at 405 nm, and nonprotein nitrogen contents showed that noncarbohydrate and nonprotein compounds were also utilized to some extent but the bacterial species involved in this degradation remain to be identified.

## KEYWORDS: Coffee dietary fiber; polysaccharides; melanoidins; fermentability; gut microbiota; enzymaticgravimetric methodology; NMR; FISH

#### INTRODUCTION

Recent studies showed that coffee beverages contain substantial amounts of soluble dietary fiber (1, 2), which is well utilized by human fecal microbiota (2). Major components of the dietary fiber fraction isolated from coffee brews are complex type II arabinogalactans and galactomannans with a low degree of substitution (2).

We recently analyzed the dietary fiber contents of various coffee brews by applying the widely accepted enzymaticgravimetric methodology (3) and showed that the contents and structural characteristics of coffee dietary fiber are influenced by various parameters such as the coffee type, the degree of roast, and the brewing procedure (4). However, it is well-known that some fiber compounds such as ethanol soluble nondigestible oligosaccharides are not determined by using the enzymatic-gravimetric methodology (5). An alternative approach has therefore been used by Díaz-Rubio and Saura-Calixto to determine dietary fiber in coffee brews (1): Following enzymatic treatments, a high molecular weight fraction was isolated by using dialysis (molecular weight cutoff (MWCO) = 12-14 kDa). After acid hydrolysis, the fiber monosaccharides were determined semispecifically by applying a colorimetric method (dinitrosalicylic acid). Although avoiding an ethanol precipitation step, this method may also lead to underestimation of the dietary fiber content because galactomannans and arabinogalactans with molecular weights <12 kDa (6) are excluded.

Although the high molecular weight fraction of coffee brews also contains considerable amounts of structurally unknown compounds derived from the Maillard reaction (melanoidins), a clear distinction between melanoidins and poly- or oligosaccharides is not possible because carbohydrates are thought to

10.1021/jf800231q CCC: \$40.75 © 2008 American Chemical Society Published on Web 06/18/2008

<sup>\*</sup> To whom correspondence should be addressed. Phone: (612) 624-1764. Fax: (612) 625-5272. E-mail: mbunzel@umn.edu.

<sup>&</sup>lt;sup>†</sup> University of Hamburg.

<sup>&</sup>lt;sup>‡</sup> German Institute of Human Nutrition.

<sup>&</sup>lt;sup>§</sup> U.S. Dairy Forage Research Center.

<sup>&</sup>quot;University of Wisconsin.

 $<sup>^{\</sup>perp}$  University of Minnesota.

be integral components of coffee melanoidin complexes (7, 8). Model melanoidins were shown to have some physiological properties similar to dietary fiber. They are mostly indigestible (9, 10), show high water holding capacities (11), and are capable of adsorbing organic molecules, such as bile acids (12). Model melanoidins were also shown to be fermented by gut microbiota, possibly affecting the microbial composition (13–15). Coffee dietary fiber isolated by the enzymatic-gravimetric procedure was previously reported to contain unknown, brown colored components (2, 4). However, the relevance of food borne melanoidins as dietary fiber still needs to be addressed.

We expected coffee brews to contain considerable amounts of high molecular weight components (melanoidins, ethanol soluble polysaccharides) that are not detected as dietary fiber using the enzymatic-gravimetric procedure but either belong to the dietary fiber complex, by definition (16), or possess fiberlike properties. Therefore, the aim of our study was to analyze the content and to determine the chemical characteristics and fermentability of this high molecular weight ethanol soluble coffee fraction.

#### MATERIALS AND METHODS

**General.** Reagents for fluorescence in situ hybridization (FISH) were of molecular biological grade. All other chemicals and solvents were of analytical grade or the highest purity available. Chemical characterization and fermentation experiments were carried out in duplicate and triplicate, respectively.

**Materials.** Light, medium, and dark roasted Arabica coffee samples (origin Colombia) were kindly provided by Tchibo Manufacturing GmbH Co. KG (Hamburg, Germany). Coffee beans were ground to an average particle size of 450  $\mu$ m.

**Coffee Preparation.** Drip coffee beverages were prepared using a standard coffee maker, No. 4 sized cone paper filters, ground coffee (50 g), and medium hard water (1 L; 69 mg/L  $Ca^{2+}$ , 3.9 mg/L  $Mg^{2+}$ , 136 mg/L  $HCO_3^{-}$ ) resulting in about 900 mL of coffee beverage. Dry weights of coffee brews were determined gravimetrically following freeze drying.

Determination and Preparation of the Ethanol Soluble High Molecular Weight Fraction. The contents of the ethanol soluble high molecular weight material from light, medium, and dark roasted Colombian coffee drip brews were analyzed gravimetrically by using the freeze-dried coffee brew extracts (1.0 g) as starting material. Coffee dietary fiber was removed according to the enzymatic-gravimetric methodology (3), as recently applied to coffee brews (2, 4) using a commercially available enzyme kit (BIOQUANT, Merck, Darmstadt, Germany). After treatments with heat-stable  $\alpha$ -amylase, protease, and amyloglucosidase, the total dietary fiber fraction was precipitated in 76% ethanol and separated by filtration (glass filter crucibles, porosity 2, pore size  $40-90 \,\mu\text{m}$ , partially filled with Celite 545). After removing ethanol by rotary evaporation at 40 °C and reduced pressure, the remaining solution was dialyzed (benzoylated cellulose, MWCO 2 kDa, Sigma-Aldrich Chemie GmbH, Steinheim, Germany) against water. The ethanol soluble high molecular weight fraction obtained after dialysis (HESF<sub>D</sub>) was freeze-dried, weighed, and subjected to total carbohydrate analysis as detailed below.

For further characterization, chemical analysis, and fermentation experiments, the ethanol soluble high molecular weight fraction was isolated on a larger scale from the medium roasted coffee brew (freeze-dried powder, 10 g) according to the described procedure but using ultrafiltration instead of dialysis. Ultrafiltration was carried out using two tangential flow devices (polyethersulfone membrane, MWCO 3 kDa, Vivaflow 50, Sartorius AG, Goettingen, Germany) connected in series and a flexible-tube pump (Masterflex L/S pump drive 7554-85 and easy-load pump head 7518-00, Cole-Parmer Instrument Co., Vernon Hills, IL). The preparation of the high molecular weight ethanol soluble fraction was performed in triplicate. The freeze-dried retentates (HESF<sub>UF</sub>, >3 kDa) were homogenized in a mortar and used as detailed below.



**Figure 1.** Fractionation of a high molecular weight ethanol soluble coffee fraction (HESF<sub>UF</sub>) using Sephadex G-50 (**A**) and Sephadex G-25 (**B**) chromatography and photo diode array detection with recording wavelengths at 405, 325, and 280 nm.

Further Fractionation of the Ethanol Soluble High Molecular Weight Fraction. HESF<sub>UF</sub> (600 mg) was further fractionated using gel filtration chromatography [L-7100 pump, L-7455 photodiode array detector equipped with a preparative flow cell (Merck/Hitachi, Darmstadt, Germany), Serva Linear II fraction collector (Serva-Technik GmbH & Co KG, Heidelberg, Germany)]. A glass column with flow adaptors (Kronlab, Sinsheim, Germany; 2.5 cm × 100 cm) was filled with Sephadex G-50 medium (Amersham Biosciences, Freiburg, Germany) swollen in water, which was also used as an eluent. HESF<sub>UF</sub> (~300 mg) dissolved in water (4.5 mL) was applied using a six-way valve and a 5 mL sample loop. The flow rate was 1.0 mL/min, and fractions were collected every 20 min and combined according to the chromatogram (Figure 1A). Fractions from two runs were pooled, freeze-dried, and subjected to chemical characterization or, in case of fraction IV, further fractionated using Sephadex G-25 medium gel filtration chromatography. Sephadex G-25 chromatography was carried out in one run using the instrumentation and column as detailed for Sephadex G-50 chromatography, water as an eluent and a flow rate of 0.5 mL/min. Fractions were collected every 10 min and combined according to the chromatogram (Figure 1B), freeze-dried, and subjected to chemical analysis as described below.

Additional fractionation of  $\text{HESF}_{\text{UF}}$  was performed by an ultrafiltration procedure. Ultrafiltration was carried out with centrifugal ultrafiltration devices (Vivaspin 20, polyethersulfone membrane, Sartorius AG, Goettingen, Germany) with MWCO of 100, 50, 10, and 3 kDa. After dissolving 200 mg of  $\text{HESF}_{\text{UF}}$  in 20 mL of water, fractions >100 kDa, 50–100 kDa, 10–50 kDa, and 3–10 kDa were obtained by stepwise ultrafiltration, freeze-dried, and weighed.

Determination of Absorbance at 405 and 280 nm. Aqueous solutions (0.1 mg/mL) of  $\text{HESF}_{\text{UF}}$  or its subfractions were measured spectrophotometrically at 280 and 405 nm (1 cm path length) using water as a blank.

Determination of Total Carbohydrates and Carbohydrate Characterization. The total carbohydrate content, calculated as the sum of anhydosugars, and the monosaccharide compositions were determined using the alditol acetate methodology: Neutral sugars were released by Saeman hydrolysis as modified by Englyst et al. (prehydrolysis with 12 M H<sub>2</sub>SO<sub>4</sub> for 5 min at room temperature, hydrolysis using 2 M

Table 1. 16S r	rRNA Targeting	Oligonucleotide	Probes	Used	for	FISH
----------------	----------------	-----------------	--------	------	-----	------

probe	sequence (5'-3')	OPD code	target organism	reference
EUB mix			total bacteria	(35)
EUB 338	GCTGCCTCCCGTAGGAGT	S-D-Bact-0338-a-A-18		(36)
EUB 785	CTACCAGGGTATCTAATCC	S-D-Bact-0785-a-A-19		(37)
EUB 927	ACCGCTTGTGCGGGCCC	S-D-Bact-0927-a-A-17		(38)
EUB 1055	CACGAGCTGACGACAGCCAT	S-D-Bact-1055-a-A-20		(37)
EUB 1088	GCTCGTTGCGGGACTTAACC	S-D-Bact-1088-a-A-20		(37)
Erec 482	GCTTCTTAGTCARGTACCG	S-*-Erec-0482-a-A-19	Clostridium coccoides-Eubacterium rectale group	(39)
Bac 303	CCAATGTGGGGGGACCTT	S-*-Bacto-0303-a-A-17	Bacteroides-Prevotella group	(40)
Bif 164	CATCCGGCATTACCACCC	S-G-Bif-0164-a-A-18	bifidobacteria	(41)

 $H_2SO_4$  for 60 min at 100 °C) (17) and analyzed as their alditol acetates (following NaBH<sub>4</sub> reduction and acetylation) by GC-FID (18). GC conditions used are detailed in Gniechwitz et al. (2).

Carbohydrate structures were also characterized by methylation analysis, which was carried out according to Nunes and Coimbra (19) with minor modifications, as previously described (2). Identification and quantification of partially methylated alditol acetates was carried out by GC-MS and GC-FID, as detailed previously (2). Molar response factors according to Sweet et al. (20) were used for quantification.

**Determination of the Total Nitrogen Content.** The total nitrogen content was analyzed after Kjeldahl digestion using the spectrophotometric procedure described by Willis et al. (21).

Amino Acid Analysis. Dried samples (1 mg) were hydrolyzed with 6 M HCl (1 mL) for 24 h at 110 °C. The acid was removed by rotary evaporation at reduced pressure and 40 °C, and the amino acids were extracted in 4% (w/v) sulfosalicylic acid (50 mL). The extracts were adjusted to 100 mL, centrifuged, and filtered. The amino acids were analyzed by HPLC and fluorescence detection after derivatization with *o*-phthaldialdehyde as described by Bytof et al. (22).

**Fecal Samples and Culture Conditions.** Fecal samples were collected from one healthy volunteer who had no previous history of gastrointestinal disorders and had not undergone antibiotic therapy within 6 months prior to the study. Fermentations were carried out using an up-scaled version (batch culture volume = 100 mL) of the procedure previously described (2). The final concentration of the HESF<sub>UF</sub> was 3.0 mg/mL. The control contained no substrate. Aliquots (30 mL) were taken after 0 h, 24 h, and 48 h of incubation and analyzed, as described below.

**Determination of Optical Density, pH and Short-Chain Fatty Acids.** Spectrophotometric determination of optical density (OD), determination of pH, and gas chromatographic determination of shortchain fatty acids (SCFA) were performed, as described previously (2).

**Preparation of Fermented Samples for Chemical Analysis.** Whereas unfermented coffee fractions were analyzed as prepared, an additional cleanup was necessary for the chemical analysis of fermented samples: Cells were removed from batch culture aliquots by centrifugation (6 min, 4 °C, 10 000g). Low molecular weight material ( $\leq$ 3 kDa) was removed from the cell-free fermented samples using the same ultrafiltration procedure, as described for the preparation of HESF<sub>UF</sub>. The retentates ( $\geq$ 3 kDa) were freeze-dried and chemically analyzed, as described before (total carbohydrate content, monosaccharide composition, total nitrogen content, total amino acid content, absorbance at 405 and 280 nm). In addition, the freeze-dried retentates were analyzed by NMR spectroscopy, as detailed below.

**NMR Spectroscopy.** Structural changes of HESF<sub>UF</sub> were monitored by NMR over the course of fermentation using two-dimensional (2D)  $^{13}C^{-1}H$  correlation spectroscopy, via the HSQC (heteronuclear single quantum coherence) experiment. Aliquots of fermented samples were taken after 0, 24, and 48 h of fermentation and prepared as described above. Samples (50 mg for 0 h-, 24 h-, and 48 h-samples, respectively) were dissolved in deuterium oxide (0.7 mL). HSQC experiments of the fermented samples were performed on a Bruker DMX-750 with a cryogenic probe whereas both the Bruker DMX-750 but also a Bruker DRX500 instrument (Rheinstetten, Germany) were used to analyze model compounds. Experiments were performed at 300 K, and chemical shifts ( $\delta$ ) were referenced to the signals of acetone, which was used as internal standard ( $\delta_{\rm H}$  2.22 ppm;  $\delta_{\rm C}$  30.89 ppm). To assign specific **Table 2.** Dry Matter, Soluble Dietary Fiber (SDF), and High Molecular Weight Ethanol Soluble Coffee Fraction (HESF<sub>D</sub>) Contents of Light, Medium, and Dark Roasted Coffee Brews and Content and Composition of Total Carbohydrates in HESF<sub>D</sub>

	light roasted	medium roasted	dark roasted
dry matter (g/100 mL)	$1.47\pm0.01$	$1.58 \pm 0.01$	$1.60\pm0.00$
SDF (mg/100 mL) <sup>a</sup>	$289 \pm 2^{b}$	$350\pm4$	$349 \pm 9$
HESF <sub>D</sub> (mg/100 mL)	$112 \pm 6^{b}$	$187 \pm 11$	$194 \pm 3$
total carbohydrates in fraction	$12.94\pm0.46$	$12.58 \pm 1.33$	$12.89 \pm 1.38$
HESF <sub>D</sub> (g/100 g)			
monosaccharide composition <sup>c</sup>			
rhamnose (mol%)	7.9	8.2	7.8
glucose (mol%)	7.7	6.6	7.9
mannose (mol%)	6.2	5.0	6.7
arabinose (mol%)	32.6	35.3	31.7
galactose (mol%)	45.6	45.0	45.8

<sup>*a*</sup> Results previously published (4). <sup>*b*</sup> Light roasted coffee showed significantly (p < 0.05) lower SDF and HESF<sub>D</sub> contents than medium or dark roasted coffee. <sup>*c*</sup> Monosaccharide composition of total carbohydrates in fraction HESF<sub>D</sub>, anhydro sugars.

signals, the following model compounds were measured under the same conditions: arabinogalactan (from larch wood), carob galactomannan (from *Ceratonia siliqua*), maltose, cellobiose, rutinose, albumin (from bovine serum), 5-*O*-caffeoylquinic acid, and L-phenylalanine.

Fluorescence in Situ Hybridization and Enumeration of Bacterial Cells. The oligonucleotide probes used in this study are listed in Table 1. Batch culture aliquots (1.0 mL) were centrifuged (5 min, 4 °C, 12000g). The pellet was resuspended in PBS (0.5 mL), added to paraformaldehyde (1.5 mL; 4% aqueous solution), and fixed for 4 h at 4 °C. Fixed samples were stored at -80 °C until further use. For microscopic sample preparation, Teflon coated 8-well-slides (Roth, Karlsruhe, Germany) with a well diameter of 6 mm were used. The fixed samples were homogenized for 1 min at full speed in a Gyroprep (UniEquip Laborgerätebau and -vertriebs GmbH, Martinsried, Germany) and diluted depending on the probe used. Tween solution (10  $\mu$ L; 0.01%) was applied to each well, and the fixed and diluted fecal samples (10  $\mu$ L) were added. The air-dried suspension was dehydrated for 3 min each in 60, 80, and 96% ethanol. In case of using probes targeting gram-positive groups of bacteria, samples were treated with lysozyme buffer pH 7.2 (14 µL; 1 mg/mL lysozyme, 100 mM Tris-HCl, 50 mM EDTA) for 10 min at 4 °C. Subsequently, the lysozyme was removed with bidistilled H<sub>2</sub>O, and the slides were air-dried and dehydrated, as described. Each well was covered with a mixture of hybridization buffer (10 µL; 0.9 M NaCl, 10 mM Tris HCl pH 7.4, 0.01% SDS) and probe solution (1  $\mu$ L; 10 pmol/ $\mu$ L in hybridization buffer), except for Bac303, which was used at 4-fold concentration in hybridization buffer with 30% (v/v) formamide. The slides were kept in a moist chamber for overnight hybridization and washed in hybridization buffer for 30 min at a temperature 2 °C higher than the one used for hybridization. The slides were air-dried and mounted with Vectashield (Vector Laboratories, Peterborough, U.K.). Samples were counted with an Axioplan2 imaging microscope (Carl Zeiss, Oberkochen, Germany), equipped with a servo-controlled microscope stage (EM14MOT, Merzhäuser, Wetzlar, Germany), a Plan-Neofluar 40× dry and 100× oil immersion objective (Carl Zeiss, Oberkochen, Germany), and the fluorescence filter set 15 (excitation: 546 nm; emission: 590 nm) for Cy3 excitation.

Table 3. Chemical Characteristics of the Ethanol Soluble High Molecular Weight Fraction (HESF<sub>UF</sub>) from a Medium Roasted Colombian Coffee Brew and of its Subfractions HESF I - IVe (Obtained by GPC)

						HESF Sub	ofraction			
		HESFUF	I	II		IVa	IVb	IVc	IVd	IVe
yield total carbohydrates <sup>a</sup> monosaccharide composition <sup>b</sup>	(g/100 g) (g/100 g)	11.46 ± 0.43	$\frac{10.7}{33.40\pm0.47}$	11.4 34.37 ± 1.79	13.0 9.16 ± 0.50	4.6 3.81 ± 0.18	20.3 10.98 ± 0.59	16.9 6.93 ± 0.13	13.9 4.07 ± 0.12	9.2 3.05 ± 0.14
rhamnose glucose mannose arabinose galactose total nitrogen nitrogen from amino acide	(mol%) (mol%) (mol%) (mol%) (g/100 g) (g/100 g)	$\begin{array}{l} 9.5 \\ 4.6 \\ 5.1 \\ 27.7 \\ 53.2 \\ 4.88 \pm 0.25 \\ 2.94 \pm 0.05 \end{array}$	$\begin{array}{c} 11.6 \\ 5.4 \\ 3.3 \\ 32.9 \\ 46.8 \\ 3.76 \pm 0.42 \\ 2.23 \pm 0.07 \end{array}$	$\begin{array}{c} 7.6 \\ 3.3 \\ 3.7 \\ 26.1 \\ 59.3 \\ 4.25 \pm 0.34 \\ 1.62 \pm 0.04 \end{array}$	$\begin{array}{c} 19.4 \\ 7.1 \\ 8.6 \\ 32.2 \\ 32.7 \\ 1.62 \pm 0.07 \\ 0.69 \pm 0.03 \end{array}$	$\begin{array}{c} 16.1 \\ 17.2 \\ 6.4 \\ 15.1 \\ 45.2 \\ 6.76 \pm 0.33 \\ 2.28 \pm 0.03 \end{array}$	$\begin{array}{c} 5.9 \\ 12.9 \\ 7.6 \\ 19.6 \\ 54.1 \\ 6.98 \pm 0.08 \\ 2.42 \pm 0.04 \end{array}$	$\begin{array}{c} 8.2 \\ 20.1 \\ 7.0 \\ 22.3 \\ 42.5 \\ 4.38 \pm 0.04 \\ 0.69 \pm 0.03 \end{array}$	$\begin{array}{c} 12.1 \\ 22.4 \\ 6.6 \\ 22.0 \\ 36.9 \\ 3.56 \pm 0.39 \\ 1.43 \pm 0.03 \end{array}$	$\begin{array}{c} 14.4 \\ 27.2 \\ 7.5 \\ 17.9 \\ 33.0 \\ 2.10  \pm  0.12 \\ 0.10  \pm  0.01 \end{array}$
"unknown" nitrogen <sup>c</sup> total amino acids <sup>d</sup> amino acid	(g/100 g) (g/100 g)	$\frac{1.94}{20.02\pm0.37}$	$\frac{1.53}{15.58\pm0.50}$	$\begin{array}{c} 2.63 \\ 11.10  \pm  0.24 \end{array}$	$\begin{array}{c} 0.93 \\ 4.55 \pm 0.12 \end{array}$	${\begin{array}{c} 4.48 \\ 15.86 \pm 0.29 \end{array}}$	4.56 17.59 ± 0.21	$\begin{array}{c} 3.69 \\ 4.75 \pm 0.20 \end{array}$	$\begin{array}{c} 2.13 \\ 9.88 \pm 0.22 \end{array}$	${\begin{array}{c} 2.00 \\ 0.68 \pm 0.07 \\ -^{e} \end{array}}$
Asx Glx Ser His Gly Thr Arg Ala Tyr Val/Met Phe Ile Leu Lys absorbance at 405 nm <sup>4</sup>	(mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (mol%) (au)	14.2 25.2 4.3 1.8 21.3 3.8 1.0 6.3 1.6 3.4 3.3 3.8 6.7 3.3 0.261	14.2 27.1 6.1 1.3 22.5 3.7 0.4 6.6 1.7 3.0 3.4 2.9 5.8 1.3 0.312	14.7 25.8 4.1 2.1 21.0 3.6 0.6 7.3 1.0 3.5 2.4 4.1 7.3 2.4 0.224	15.6 25.5 4.5 1.7 22.2 4.1 1.7 6.9 0.3 3.3 2.5 3.0 6.2 2.3 0.073	13.7 24.5 3.1 1.7 24.3 3.9 0.5 5.9 2.8 2.9 4.7 4.0 6.7 1.3 0.273	16.4 31.0 3.5 1.1 16.5 4.6 0.5 7.3 1.0 3.2 2.7 3.7 7.2 1.5 0.172	16.4 26.0 4.3 1.9 20.0 3.6 0.3 6.2 0.8 2.8 3.3 3.8 7.1 3.6 0.138	18.3 25.6 4.5 2.6 16.8 5.2 0.8 6.5 0.5 3.5 2.5 4.0 6.9 2.5 0.277	0.300
absorbance at 280 nm <sup>f</sup>	(au)	0.893	0.697	0.643	0.202	0.732	0.631	0.553	0.959	1.107

<sup>a</sup> Anhydro sugars. <sup>b</sup> Molar percentages of total carbohydrate or total amino acid contents, respectively. <sup>c</sup> Nitrogen deriving from structurally unknown compounds (calculated). <sup>d</sup> Anhydro amino acids. <sup>e</sup> Not listed (due to low total amino acid content only three amino acids were measurable). <sup>f</sup> Aqueous solutions (0.1 mg/mL) d = 1 cm.

**Table 4.** Comparison of Selected Structural Characteristics ofArabinogalactans Present in the Ethanol Soluble High Molecular WeightFraction (HESFUF) and the Soluble Dietary Fiber Fraction (SDF) from aMedium Roasted Colombian Coffee Brew

glycosidic-linkage ratio	HESFUF	SDF <sup>a</sup>
3-Galp/6-Galp	1.9	2.8
(3-Galp + 6-Galp)/3, 6-Galp	2.7	1.9
T-Araf/3,6-Galp	3.0	1.1
T-Araf/5-Araf	1.5	3.6

<sup>a</sup> Results previously published (4).

#### **RESULTS AND DISCUSSION**

Contents and Characteristics of the Ethanol Soluble High Molecular Weight Coffee Fraction (HESF<sub>D/UF</sub>). Using dialysis (MWCO 2 kDa), drip brews from light, medium, and dark roasted Colombian coffees were shown to contain 112–194 mg/ 100 mL of high molecular weight compounds that are not detected as dietary fiber (HESF<sub>D</sub>) using the widely accepted enzymatic-gravimetric procedure (3) (**Table 2**). These fractions contributed to 8-12% of the coffee brew dry matter. There was a significant increase in the HESF<sub>D</sub> contents from light to medium roasted coffee but no further increase for the dark roasted brew. Similar effects of the roasting degree were observed upon analyzing the soluble dietary fiber (SDF) contents of coffee beverages (4) (**Table 2**). However, only about 13% of this high molecular weight fraction were carbohydrates and therefore dietary fiber by definition (*16*). The monosaccharide composition of  $\text{HESF}_{\text{D}}$  was comparable for all degrees of roast (**Table 2**) and revealed that ethanol soluble carbohydrates were mainly arabinogalactans. This is in accordance with results reported by Bekedam et al. (23): arabinose and galactose made up about 80 mol% of carbohydrates present in a coffee fraction that was soluble in 80% ethanol. Compared to SDF arabinogalactans (4), HESF<sub>D</sub> carbohydrates were characterized by lower galactose/arabinose ratios (1.3–1.4 vs 2.1–3.2).

The high molecular weight ethanol soluble fraction was isolated on a larger scale from the medium roasted coffee brew using ultrafiltration instead of dialysis (HESF<sub>UF</sub>). The average yield of HESF<sub>UF</sub> ( $\sim$ 11% of the coffee brew dry matter) was similar to the yield of HESF<sub>D</sub> ( $\sim$ 12%).

Further analysis and fermentation experiments were carried out with HESF<sub>UF</sub>. Its monosaccharide composition was slightly different from HESF<sub>D</sub>. However, the main monosaccharide residues were also arabinose and galactose, accounting for ~80% of the total carbohydrates (**Table 3**). Apart from the already mentioned galactose/arabinose ratio, arabinogalactans from HESF<sub>UF</sub> and SDF varied also in other structural characteristics as analyzed by methylation analysis (**Table 4**): the HESF<sub>UF</sub> arabinogalactans were less branched as indicated by a higher (3-Gal*p* + 6-Gal*p*)/3,6-Gal*p* ratio (2.7 vs 1.9). In addition, HESF<sub>UF</sub> arabinogalactans were characterized by lower 3-Gal*p*/ 6-Gal*p* (1.9 vs 2.8) and T-Araf/5-Araf ratios (1.5 vs 3.6). Furthermore, HESF<sub>UF</sub> showed a much higher ratio of T-Araf/ 3,6-Gal*p* than SDF (3.0 vs 1.1). Nunes and Coimbra also

Table 5. Production of SCFA after 24 and 48 h of Fermentation of a High Molecular Weight Ethanol Soluble Coffee Fraction (HESF<sub>UF</sub>) Compared to the Control

	acetate (mM)		propionate (mM)		butyrate (mM)		total (mM)	
	24 h	48 h	24 h	48 h	24 h	48 h	24 h	48 h
HESF <sub>UF</sub> control	3.70 ± 0.70 1.12	$\begin{array}{c} 4.63 \pm 0.01 \\ 1.55 \end{array}$	1.17 ± 0.99 0.06	$\begin{array}{c} 1.89\pm0.03\\ 0.05\end{array}$	$\begin{array}{c} 0.59\pm0.09\\ 0.04\end{array}$	$\begin{array}{c} 0.85\pm0.02\\ 0.03\end{array}$	$\begin{array}{c} 5.47 \pm 0.97 \\ 1.22 \end{array}$	$\begin{array}{c} 7.36\pm0.05\\ 1.63\end{array}$

Table 6. Total Carbohydrate, Total Nitrogen, and Amino Acid Nitrogen Contents as Analyzed after 0, 24, and 48 h of Fermentation of a High Molecular Weight Ethanol Soluble Coffee Fraction (HESF<sub>UF</sub>) by Human Fecal Microbiota

		0 h	01 h	10 h
		0 11	24 11	40 11
yield	(g/100 g)	72.9	55.8	61.9
total nitrogen	(g/100 g)	$3.55\pm0.06$	$2.55\pm0.01$	$1.96\pm0.02$
total amino acids <sup>a</sup>	(g/100 g)	$12.41\pm0.67$	$10.39\pm0.58$	$7.91\pm0.86$
nitrogen from amino	(g/100 g)	$1.85\pm0.09$	$1.53\pm0.07$	$1.17\pm0.11$
acids				
"unknown" nitrogen <sup>b</sup>	(g/100 g)	1.70	1.02	0.79
total carbohydrates <sup>c</sup>	(g/100 g)	$14.44\pm0.28$	$6.57\pm0.13$	$4.88\pm0.10$
rhamnose <sup>c</sup>	(g/100 g)	$1.21\pm0.02$	$0.57\pm0.03$	$0.19\pm0.02$
glucose <sup>c</sup>	(g/100 g)	$0.36\pm0.03$	$0.23\pm0.02$	$1.28\pm0.00$
mannose <sup>c</sup>	(g/100 g)	$0.51\pm0.04$	$0.27\pm0.01$	$0.21\pm0.00$
arabinose <sup>c</sup>	(g/100 g)	$4.34\pm0.18$	$2.50\pm0.12$	$1.10\pm0.09$
galactose <sup>c</sup>	(g/100 g)	$\textbf{8.02} \pm \textbf{0.21}$	$\textbf{3.00} \pm \textbf{0.02}$	$2.10\pm0.04$

<sup>a</sup> Anhydro amino acids. <sup>b</sup> Nitrogen deriving from structurally unknown compounds (calculated). <sup>c</sup> Anhydro sugars.

observed high T-Araf/3,6-Galp-ratios (2.1–2.6) in melanoidins they isolated from the 75% ethanol soluble fractions of roasted Brazilian coffees (8).

The differences in arabinogalactan structures in HESF<sub>UF</sub> and SDF may partially explain their different ethanol solubilities. However, ethanol solubility may also be affected by linkages to structurally unknown compounds (e.g., Maillard reaction products (MRP)). The presence of high amounts of MRPs in HESF<sub>UF</sub> was indicated by a strong brown color and UV-activity (aqueous solution, 0.1 mg/mL, absorbances at 405 and 280 nm; **Table 3**), which was 3–4-fold higher than observed for the SDF fraction (aqueous solution, 0.25 mg/mL, 0.206 au and 0.509 au for 405 and 280 nm, respectively) (4). In addition, HESF<sub>UF</sub> contained 4.88 g/100 g nitrogen, only about 60% of which derived for 31.5% of the HESF, again indicating the importance of structurally unknown MRPs for this ethanol soluble high molecular weight material.

Stepwise ultrafiltration of HESFUF showed that its constituents substantially differed in their molecular weights, most of the compounds being found in fractions >100 kDa (40%) or 3-10kDa (53%). Fractions 10-50 kDa and 50-100 kDa accounted only for 5 and 2%, respectively. The broad molecular weight range of compounds present in HESF<sub>UF</sub> is also reflected in the chromatograms obtained by gel filtration using Sephadex G-50 (Figure 1A). The majority of compounds eluted after  $\sim$ 440 min, as indicated by the elution profile measured at 405 nm, 325, or 280 nm and by the yields of the fractions (Table 3). Because the structures and therefore the conformations of compounds in HESF<sub>UF</sub> are essentially unknown, a calibration of the gelfiltration chromatography with comparable standard substances is not possible. However, a 1 kDa dextran standard eluted after 442 min, indicating a rather low molecular weight of compounds of fraction IV. The yield of fraction IV was roughly comparable to the proportion of the 3-10 kDa fraction obtained by ultrafiltration (65% vs 53%). However, it is possible that secondary interactions between coffee melanoidins and the gel matrix occur, leading to an underestimation of molecular weights. Further fractionation of fraction IV using Sephadex G-25 gel chromatography indicated the complexity of the material in the lower high molecular weight range (**Figure 1B**).

Chemical analysis of the fractions obtained by gel-filtration chromatography revealed that the early eluted components (subfractions I and II) had the highest carbohydrate contents (33 and 34%, respectively) (Table 3). It may be concluded that longer polysaccharide chains are involved in melanoidin structures present in fractions I and II, compared to fractions III to IVe, leading to higher molecular weights. However, the gelfiltration fractions rather are complex mixtures than pure compounds. Therefore, it cannot be ruled out that they also contain ethanol-soluble carbohydrates, which are not part of melanoidin structures. Arabinose and galactose proportions (Table 3) showed that carbohydrate components of all HESF<sub>UF</sub> subfractions are mainly arabinogalactans. Arabinose/galactose ratios ranged from 0.3 (subfraction IVa) to 1.0 (subfraction III) and follow no trend regarding the assumed molecular weights. Surprisingly high glucose proportions were measured for subfractions IVa to IVe. However, we cannot fully rule out that compounds other than glucose, which may have been released or formed from MRP during acid hydrolysis coeluted with glucitolperacetate, thus leading to an overestimation of the glucose content.

HESF<sub>UF</sub> subfractions largely differed in their total nitrogen (1.6-7.0%) and total amino acid contents (0.7-17.6%), but no correlations to molecular weight or absorbances at 405 and 280 nm were distinct.

To sum up, the data show that the high molecular weight coffee fraction, which is not determined as dietary fiber, is composed of a broad spectrum of compounds that differ in their molecular weights and chemical compositions although some coarse similarities (e.g., in the carbohydrate composition) exist. Because of the high content of chemically unknown substances, as well as the distinct brown color and UV-activity, it may be concluded that this coffee fraction is mainly composed of melanoidins with low carbohydrate and protein contents and this seems especially true for those subfractions with assumed lower molecular weights. The structures of coffee melanoidins, like those of food melanoidins in general, are still largely unknown. However, oligo- or polysaccharides and/or peptides/ proteins as well as phenolic compounds are proposed as components of these complex structures in coffee deriving from the Maillard reaction (7, 8, 23).

**Fermentability of the Ethanol Soluble High Molecular Weight Coffee Fraction.** Although HESF is not determined as dietary fiber (enzymatic-gravimetric method), it is partially composed of nondigestible oligo- or polysaccharides that are however most likely incorporated into melanoidin structures. The fermentability of HESF<sub>UF</sub> was investigated by in vitro fermentation experiments with human fecal bacteria. Increasing OD over the course of fermentation indicated that HESF<sub>UF</sub> was utilized by fecal microbiota in vitro. However, this increase was less distinct (ca. 0.3 units) when compared to fermentation of SDF (ca. 0.45 units) (2) despite fermenting twice the amount



Figure 2. Anomeric region of HSQC spectra for a high molecular weight ethanol soluble coffee fraction (HESF<sub>UF</sub>) (**A**) after 0, 24 and 48 h of fermentation, rutinose (**B**), larch arabinogalactan (**C**), carob galactomannan (**D**), and cellobiose (**E**). The color code is as follows: yellow, rhamnopyranosyl residues; purple, mannopyranosyl residues; magenta, galactopyranosyl residues; green, glucopyranosyl residues; blue, arabinofuranosyl residues. For signal numbers refer to text and **Table 7**.

of  $\text{HESF}_{\text{UF}}$ . Because  $\text{HESF}_{\text{UF}}$  contained considerably less carbohydrates than SDF, these results indicate that noncarbo-

hydrate components of  $\text{HESF}_{\text{UF}}$  are less fermentable than coffee carbohydrates.

Table 7. Assignment of Characteristic Carbohydrate Correlations Present in HSQC Spectra of HESFUF (Presented in Figure 2)

correlation ( $\partial H^1 / \partial C^{13}$ )	No. (Figure 2)	model compound for comparison	assignment	reference
4.84/100.39 ppm	1	rutinose	$\alpha$ -linked T-Rhap	-
4.84/100.39 ppm	2	carob galactomannan	$\beta$ -linked 4-Man $p$	(25)
4.50/103.27 ppm	3	cellobiose	$\beta$ -linked nonreducing Glcp	-
4.77-4.34/103.2-104.8 ppm	4	larch arabinogalactan	$\beta$ -linked Gal $p^a$	(26)
5.15/107.5 ppm	5	-	not assigned	-
5.07/108.08 ppm	6	-	$\alpha$ -linked 5-Arat <sup>b</sup>	-
5.23/109.82 ppm	7	larch arabinogalactan	α-linked T-Arat <sup>b</sup>	(27, 28)
5.42/108.9 ppm	8	-	not assigned	-

<sup>a</sup> A differentiation between correlations from T-Galp, 3-Galp, 6-Galp and 3,6-Galp was not attempted in this study. <sup>b</sup> Tentatively assigned (see text).



Figure 3. HSQC spectra of a high molecular weight ethanol soluble coffee fraction (HESF<sub>UF</sub>) after 0, 24, and 48 h of fermentation displayed at high (A) and low (B) contour levels.

After 24 h of fermentation, acetate, propionate, and butyrate were produced in low concentrations in a molar ratio of  $\sim$ 68: 21:11 (**Table 5**), demonstrating higher proportions of propionate and butyrate than recently reported for the fermentation of coffee SDF (81:15:4) (2). Fermentation studies using larch arabinogalactan or guar as substrates showed that the microbial degradation of galactomannans leads to the production of higher molar

proportions of acetate than the degradation of arabinogalactans (24). Since SDF polysaccharides were characterized by higher proportions of galactomannans than HESF<sub>UF</sub> carbohydrates ( $\sim$ 50 mol% vs  $\sim$ 5 mol% mannose), this is a possible explanation for the differences in molar ratios of SCFA (although arabinogalactans from larch and galactomannans from guar are not completely comparable to those from coffee).



Figure 4. Ferulic acid (left) and caffeic acid (right); carbons and protons that were thought to show characteristic correlation signals in the HSQC spectra are numbered.

Chemical analysis showed that HESF<sub>UF</sub> polysaccharides were well degraded: after 24 h of fermentation, the remaining high molecular weight material contained 55% and after 48 h 66% less carbohydrates than the 0 h-sample (Table 6). Still, compared to the fermentation of coffee SDF, where about 91% of SDF carbohydrates were degraded after 24 h of fermentation (2), the carbohydrate degradation was less extensive. However, we also showed that coffee arabinogalactans were less utilized than galactomannans (2, 6). In addition, HESF<sub>UF</sub> arabinogalactans are possibly less degradable than SDF arabinogalactans because they show lower T-Araf/5-Araf and 3-Galp/6-Galp ratios. Also, it can not be ruled out that the high proportion of MRPs in coffee HESF<sub>UF</sub> negatively affect carbohydrate degradation. Finally, the lower bacterial growth, which was assumed to be due to limited substrate availability, may have resulted in a slower utilization of carbohydrates.

It was also shown that the total content of amino acids decreased over the course of fermentation (Table 6). However, no significant amounts of branched SCFA, known as microbial products from protein utilization, were measured after 24 or 48 h of fermentation (data not shown). In addition to polysaccharides and amino acids, MRPs were also utilized by fecal microbiota, as indicated by the decreasing contents of "unknown", non-amino acid nitrogen (Table 6). Despite the highly complex nature of this high molecular weight coffee fraction, HESF<sub>UF</sub> fermentation was successfully monitored by using 2D-NMR <sup>13</sup>C<sup>-1</sup>H correlation experiments. To ensure comparability, the same sample amounts (residual high molecular weight material after 0, 24, and 48 h; 50 mg of each) were analyzed using the same conditions (NMR instrument, number of scans, temperature, solvent), and spectra were compared at similar contour levels. The relative decrease in the signals observed in the carbohydrate anomeric region from 0 to 24 h and 48 h of fermentation is shown in Figure 2A. By using carbohydrate model compounds and data available from the literature (25-28), most signals in the anomeric region could be assigned to <sup>1</sup>H/ <sup>13</sup>C correlations of specific monosaccharide residues (summarized in **Table 7**;). Correlations **6** (5.07 ppm/108.08 ppm) and 7 (5.23 ppm/109.82 ppm) were only tentatively assigned to  $\alpha$ -linked 5-Araf and to T-Araf, respectively. The C-1 chemical shift of T-Araf from larch arabinogalactans was reported to be around 110 ppm (27, 28), whereas 5-Araf is not known as larch arabinogalactan constituent (27-29). This is in accordance with our larch arabinogalactan spectrum (Figure 2C). However, for the anomeric carbons of 5-Araf and T-Araf from arabinans, chemical shifts of  $\sim 108$  ppm were reported (30-32). Thus, signal 6 may also include the anomeric T-Araf correlation signal.

NMR spectroscopy confirmed that the carbohydrates in  $\text{HESF}_{\text{UF}}$  were readily utilized by fecal microbiota. Figure 3 shows the HSQC spectra of 0 h-, 24 h-, and 48 h-samples at high (**A**, left) and low (**B**, right) contour levels. Considering the spectra at higher contour levels, the intensities of the



**Figure 5.** Total cell counts during fermentation of a high molecular weight ethanol soluble coffee fraction (HESF<sub>UF</sub>) (after 0, 24, and 48 h) as analyzed by fluorescence in situ hybridization. Control = medium and bacteria without substrate.



**Figure 6.** Cell counts of different bacterial groups during fermentation of a high molecular weight ethanol soluble coffee fraction (HESF<sub>UF</sub>) (after 0, 24, and 48 h) as analyzed by fluorescence in situ hybridization: (**A**) *Eubacterium rectale-Clostridium coccoides* group; (**B**) bifidobacteria; (**C**) *Bacteroides-Prevotella* group. Control = medium and bacteria without substrate.

correlations marked by a rectangle decrease over time. All carbohydrate correlations (with the exception of the methyl correlation from rhamnose) are located in this part of the spectrum. At low contour levels (**Figure 3B**), the spectra of the fermented samples revealed no obvious decrease in the intensity of correlations, which are located in the region of phenolic, aromatic, or olefinic proton/carbon correlations (denoted in a circle). Phenolic compounds are thought to be incorporated into coffee melanoidins (7, 8, 23, 33), and chlorogenic acids are often proposed to play a role in coffee melanoidin formation. However, characteristic correlations from intact caffeic or ferulic acid moieties (**Figure 4**) of chlorogenic

acids were absent in the HSQC<sub>UF</sub> (0 h-sample): no correlations assigned to H-7/C-7 (7.49 ppm/146.86 ppm) or H-6/C-6 (7.01 ppm/123.40 ppm) of caffeic acid (spectrum of 5-O-caffeoylchinic acid not shown) were present even at contour levels close to the noise. The characteristic correlations deriving from H-7/ C-7 (~7.6-7.7 ppm/147 ppm) and H-2/C-2 (~7.1-7.2 ppm/ 112 ppm) of ferulic acid (34) were also not present in the HSQC spectrum of HESF<sub>UF</sub>. Absence of these correlations implies that intact caffeic or ferulic acid moieties are not (or insignificantly) incorporated into the melanoidins of the ethanol soluble high molecular weight coffee fraction. Hence, the circled correlations (Figure 3A) are suggested to derive from other compounds. The spectra at low contour levels (Figure 3B) also revealed that, in addition to polysaccharides, other yet unknown compounds (correlations delineated by a rectangle) are utilized by fecal bacteria. In this region, no correlations of proteins or carbohydrates are expected, as confirmed by the HSQC spectra of carbohydrates (larch arabinogalactan, carob galactomannan, and cellobiose) and bovine serum albumin (spectrum not shown) and comparison with data from the literature. This finding indicates a degradation or modification of unknown noncarbohydrate and nonprotein compounds deriving from the Maillard reaction. In the marked region (rectangle), correlations of noncarbohydrate and nonprotein carbon atoms bound to a proton and a heteroatom are expected (e.g., HC-O or HC-N), but no specific suggestions are possible at this point. To sum up, the results of our in vitro study indicate that noncarbohydrate and nonprotein moieties of coffee melanoidin complexes are fermented in the colon as well. However, whether or not the high molecular weight coffee fraction, which is not measured as dietary fiber using the enzymatic gravimetric procedure, adds to the dietary fiber content of coffee beverages beyond its carbohydrate content remains to be addressed.

Effects on the Composition of the Fecal Microbiota. FISH analysis was carried out to monitor changes in the proportion of dominant bacterial groups during degradation of HESF<sub>UF</sub>. The proportions of three dominant bacterial groups relative to the total cell counts at 0, 24, and 48 h (Figure 5) were compared. Whereas the proportions of members of the Eubacterium rectale-Clostridium coccoides group and of bifidobacteria decreased (Figure 6A and 6B, respectively), the proportion of members of the Bacteroides-Prevotella group was fairly stable over 48 h of incubation (Figure 6C). Thus, members of the Bacteroides-Prevotella group were possibly capable of partly utilizing HESF<sub>UF</sub> as a substrate. Because bacteroides are wellknown for their ability to degrade various polysaccharides including polysaccharides from coffee brews (2), we assume that members of this group are responsible for HESF<sub>UF</sub> polysaccharide degradation. The small increase in the total cell count was most likely due to substrate limitations, owing to the low carbohydrate content of HESFUF. However, the polysaccharide composition with arabinogalactans as dominating carbohydrates might also have some impact.

A large proportion of the microbiota remained unidentified and further investigations are needed to reveal which bacterial groups are involved in the degradation of these high molecular noncarbohydrate and nonprotein coffee compounds.

#### **ABBREVIATIONS USED**

5-Araf,  $(1\rightarrow 5)$ -linked arabinofuranosyl residues; 3-Galp,  $(1\rightarrow 3)$ -linked galactopyranosyl residues; 6-Galp,  $(1\rightarrow 6)$ -linked galactopyranosyl residues; 3,6-Galp,  $(1\rightarrow 3,6)$ -linked galactopyranosyl residues; FISH, fluorescence in situ hybridization; HESF<sub>D</sub> and HESF<sub>UF</sub>, high molecular weight ethanol soluble

coffee fraction, obtained by dialysis and ultrafiltration, respectively; HSQC, heteronuclear single quantum coherence; MRP, Maillard reaction products; MWCO, molecular weight cutoff; OD, optical density; PBS, phosphate buffered saline; SDF, soluble dietary fiber; T-Ara*f*, terminally linked arabinofuranosyl residues; T-Gal*p*, terminally linked galactopyranosyl residues; T-Rha*p*, terminally linked rhamnopyranosyl residues; TRIS, tris(hydroxymethyl)aminomethane.

#### ACKNOWLEDGMENT

The authors want to thank Prof. Dirk Selmar and Dr. Gerhard Bytof for experimental support with amino acid analysis.

## LITERATURE CITED

- Díaz-Rubio, M. E.; Saura-Calixto, F. Dietary fiber in brewed coffee. <u>J. Agric. Food Chem.</u> 2007, 55, 1999–2003.
- (2) Gniechwitz, D.; Reichardt, N.; Blaut, M.; Steinhart, H.; Bunzel, M. Dietary fiber from coffee beverage-degradation by human fecal microbiota. *J. Agric. Food Chem.* 2007, *55*, 6989–6996.
- (3) Lee, S. C.; Prosky, L.; deVries, J. W. Determination of total, soluble, and insoluble dietary fiber in foods-Enzymatic-gravimetric method, MES-TRIS Buffer: collaborative study. <u>J. AOAC</u> <u>Int.</u> 1992, 75, 395–416.
- (4) Gniechwitz, D.; Brückel, B.; Reichardt, N.; Blaut, M.; Steinhart, H.; Bunzel, M. Coffee dietary fiber contents and structural characteristics as influenced by coffee type, technological and brewing procedures. <u>J. Agric. Food Chem.</u> 2007, 55, 11027– 11034.
- (5) McCleary, B. V. Dietary fibre analysis. <u>Proc. Nutr. Soc</u>. 2003, 62, 3–9.
- (6) Gniechwitz, D.; Reichardt, N.; Blaut, M.; Steinhart, H.; Bunzel, M. High molecular weight components from coffee beveragesdegradation by human intestinal bacteria In Proceedings of the 21st International Conference on Coffee Science (Montpellier); Association for Science and Information on Coffee: Paris, France, 2006; pp 73–79.
- (7) Bekedam, E. K.; De Laat, M. P. F. C.; Schols, H. A.; Van Boekel, M. A. J. S.; Smit, G. Arabinogalactan proteins are incorporated in negatively charged coffee brew melanoidins. <u>J. Agric. Food</u> <u>Chem.</u> 2007, 55, 761–768.
- (8) Nunes, F. M.; Coimbra, M. A. Melanoidins from coffee infusions. Fractionation, chemical characterization, and effect of the degree of roast. *J. Agric. Food Chem.* 2007, *55*, 3967–3977.
- (9) Nair, B. M.; Öste, R.; Asp, N. G.; Pernemalm, P.-A. Absorption and distribution of a C(14)-glucose lysine reaction mixture in the rat. *Prog. Food Nutr. Sci.* **1981**, *5*, 217–222.
- (10) Finot, P. A.; Magnenat, E. Metabolic transit of early and advanced Maillard products In *Progress in Food and Nutrition Science*; Erikkson, C., Ed.; Pergamon Press: Oxford, U.K., 1981; pp 193– 207.
- (11) O'Brien, J.; Walker, R. Toxicological effects of dietary Maillard reaction products in the rat. <u>Food Chem. Toxicol</u>. **1988**, 26, 775– 783.
- (12) Ragot, F. I. J.; Russell, G. F.; Schneeman, B. O. Effect of Maillard reaction product on bile acid binding, plasma and hepatic lipids, and weight of gastrointestinal organs. *J. Agric. Food Chem.* **1992**, 40, 1634–1640.
- (13) Ames, J. M.; Wynne, A.; Hofmann, A.; Plos, S.; Gibson, G. R. The effect of a model melanoidin mixture on faecal bacterial populations in vitro. <u>Br. J. Nutr.</u> 1999, 82, 489–495.
- (14) Dell'Aquila, C.; Ames, J. M.; Gibson, G. R.; Wynne, A. G. Fermentation of heated gluten systems by gut microflora. <u>*Eur.*</u> <u>Food Res. Technol.</u> 2003, 217, 382–386.
- (15) Horikoshi, M.; Ohmura, M.; Gomyo, T.; Kuwabara, Y.; Ueda, S. Effects of browning products on the intestinal microflora of the rat In *Progress in Food and Nutrition Science*; Erikkson, C., Ed.; Pergamon Press: Oxford, U.K., 1981; pp 223–228.

- (16) DeVries, J. W. The definition of dietary fiber. <u>Cereal Foods World</u> 2001, 46, 112–126.
- (17) Englyst, H. N.; Quigley, M. E.; Geoffrey, J. H. Determination of dietary fibre as non-starch polysaccharides with gas-liquid chromatographic, high performance liquid chromatographic or spectrophotometric measurement of constituent sugars. <u>Analyst</u> 1994, 119, 1497–1509.
- (18) Blakeney, A. B.; Harris, P. J.; Henry, R. J.; Stone, B. A. A simple and rapid preparation of alditol acetates for monosaccharide analysis. *Carbohvdr. Res.* **1983**, *113*, 291–299.
- (19) Nunes, F. M.; Coimbra, M. A. Chemical characterization of galactomannans and arabinogalactans from two arabica coffee infusions as affected by the degree of roast. *J. Agric. Food Chem.* 2002, *50*, 1429–1434.
- (20) Sweet, D. P.; Shapiro, R. H.; Albersheim, P. Quantitative analysis by various g.l.c. response-factor theories for partially methylated and partially ethylated alditol acetates. *Carbohvdr. Res.* 1975, 40, 217–225.
- (21) Willis, R. B.; Montgomery, M. E.; Allen, P. R. Improved method for manual, colorimetric determination of total Kjeldahl nitrogen using salicylate. *J. Agric. Food Chem.* **1996**, *44*, 1804–1807.
- (22) Bytof, G.; Knopp, S.-E.; Schieberle, P.; Teutsch, I.; Selmar, D. Influence of processing on the generation of γ-aminobutyric acid in green coffee beans. *Eur. Food Res. Technol.* 2005, 220, 245– 250.
- (23) Bekedam, E. K.; Schols, H. A.; van Boekel, M. A. J. S.; Smit, G. High molecular weight melanoidins from coffee brew. <u>J. Agric.</u> <u>Food Chem.</u> 2006, 54, 7658–7666.
- (24) Robinson, R.; Causey, J.; Slavin, J. L. Nutritional benefits of larch arabinogalactans In Advanced Dietary Fibre Technology; Prosky, L., Ed.; Blackwell Science: Oxford, U.K., 2001; pp 443–451.
- (25) Navarini, L.; Gilli, R.; Gombac, V.; Abatangelo, A.; Bosco, M.; Toffanin, R. Polysaccharides from hot water extracts of roasted coffea arabica beans: isolation and characterization. <u>*Carbohydr.*</u> *Polym.* **1999**, *40*, 71–81.
- (26) Karacsonyi, S.; Kovacik, V.; Alföldi, J.; Kubackova, M. Chemical and <sup>13</sup>C-N.M.R. studies of an arabinogalactan from *Larix sibirica* L. *Carbohydr. Res.* **1984**, *134*, 265–274.
- (27) Willför, S.; Sjöholm, R.; Laine, C.; Holmbom, B. Structural features of water-soluble arabinogalactans from Norway spruce and Scots pine heartwood. <u>*Wood Sci. Technol.*</u> 2002, *36*, 101– 110.
- (28) Manley-Harris, M. Structural studies by NMR spectroscopy of the major oligomers from alkali-degraded arabinogalactan from *Larix occidentalis*. <u>*Carbohydr. Polym.*</u> 1997, 34, 243–249.
- (29) Ponder, G. R.; Richards, G. N. Arabinogalactan from Western larch, part III: alkaline degradation revisited, with novel conclusions on molecular structure. <u>*Carbohydr. Polym.*</u> 1997, 34, 251– 261.
- (30) Ryden, P.; Colquhoun, I. J.; Selvendran, R. R. Investigation of structural features of the pectic polysaccharides of onion by 13C-N.M.R. spectroscopy. *Carbohvdr. Res.* **1989**, *185*, 233–237.

- (31) Joseleau, J.-P.; Chambat, G.; Lanvers, M. Arabinans from the roots of horsebean (*Vicia faba*). *Carbohvdr. Res.* 1983, 122, 107–113.
- (32) Colquhoun, I. J.; Ralet, M. C.; Thibault, J. F.; Faulds, C. B.; Williamson, G. Structure identification of feruloylated oligosaccharides from sugar-beet pulp by NMR-spectroscopy. <u>*Carbohydr.*</u> <u>*Res.*</u> 1994, 263, 243–256.
- (33) Adams, A.; Borrelli, R. C.; Fogliano, V.; de Kimpe, N. Thermal degradation studies of food melanoidins. <u>J. Agric. Food Chem.</u> 2005, 53, 4136–4142.
- (34) Bunzel, M.; Ralph, J.; Steinhart, H. Association of non-starch polysaccharides and ferulic acid in grain amaranth (*Amaranthus caudatus* L.) dietary fiber. <u>Mol. Nutr. Food Res</u>. 2005, 49, 551– 559.
- (35) Kleessen, B.; Hartmann, L.; Blaut, M. Oligofructose and longchain inulin: influence on the gut microbial ecology of rats associated with a human faecal flora. <u>Br. J. Nutr</u>. 2001, 86, 291– 300.
- (36) Amann, R. I.; Binder, B. J.; Olson, R. J.; Chisholm, S. W.; Devereux, R.; Stahl, D. A. Combination of 16S rRNA-targeted oligonucleotide probes with flow cytometry for analyzing mixed microbial populations. <u>*Appl. Environ. Microbiol.*</u> 1990, 56, 1919– 1925.
- (37) Lee, S. H. Use of multiple 16S rRNA-targeted fluorescent probes to increase signal strength and measure cellular RNA from natural planktonic bacteria. *Mar. Ecol.: Prog. Ser.* **1993**, *101*, 193–201.
- (38) Giovannoni, S. J.; DeLong, E. F.; Olsen, G. J.; Pace, N. R. Phylogenetic group-specific oligodeoxynucleotide probes for identification of single microbial cells. *J. Bacteriol.* **1988**, *170*, 720–726.
- (39) Franks, A. H.; Harmsen, H. J.; Raangs, G. C.; Jansen, G. J.; Schut, F.; Welling, G. W. Variations of bacterial populations in human feces measured by fluorescent in situ hybridization with groupspecific 16S rRNA-targeted oligonucleotide probes. *Appl. Environ. Microbiol.* **1998**, *64*, 3336–3345.
- (40) Manz, W.; Amann, R.; Ludwig, W.; Vancanneyt, M.; Schleifer, K. H. Application of a suite of 16S rRNA-specific oligonucleotide probes designed to investigate bacteria of the phylum cytophagaflavobacter-bacteroides in the natural environment. <u>*Microbiology*</u> **1996**, *142*, 1097–1106.
- (41) Langendijk, P. S.; Schut, F.; Jansen, G. J.; Raangs, G. C.; Kamphuis, G. R.; Wilkinson, M. H. F.; Welling, G. W. Quantitative fluorescence in situ hybridization of Bifidobacterium spp. with genus-specific 16S rRNA-targeted probes and its application in fecal samples. *Appl. Environ. Microbiol.* **1995**, *61*, 3069–3075.

Received for review January 22, 2008. Revised manuscript received April 21, 2008. Accepted April 27, 2008. This research project was supported by the German Ministry of Economics and Technology (via AiF) and the FEI (Forschungskreis der Ernährungsindustrie e.V., Bonn). Project AiF 14085 BG.

JF800231Q